## VOLCANIC RECORDS AND DATING OF AN UPPER HALF OF THE H15 ICE CORE FROM MIZUHO PLATEAU, EAST ANTARCTICA

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Abstract: Two closely-spaced peaks of electrical conductivity were foundat depths between 45 and 50 m of the 120-m long ice core drilled at site H15 in Antarctica by the 32nd Japanese Antarctic Research Expedition in 1991. Chemical analysis of the core containing these peaks revealed that the ice layers were characterized by high acidity and high sulfate concentration suggesting volcanic signals. In order to identify these characteristics, the core was dated using two methods: (1) counting the number of high electrical conductivity peaks resulting from seasonal variations of  $SO_4^{2-}$  and  $NO_3^{-}$ , and (2) an empirical model of firn densification. The dating results suggest that the high conductivity is related to deposition of acidic aerosols from the volcanic eruptions of Tambora in 1815 A. D. and of an unknown volcano in 1809 A. D.

#### 1. Introduction

Ice sheet of Arctic and Antactic regions contains atmospheric aerosols of volcanic, marine and artificial origins which deposit as dry fall-out and wash-out. They occur as aggregates in snow particles. Ice cores drilled from ice sheet, therefore, include records of past volcanic activities, climatic and environmental changes.

A huge amount of volcanic gases and dust is emitted to the atmosphere at the time of explosive volcanic eruption. These volcanic products, especially volcanic sulfur gases, are oxidized to form aerosols of sulfuric acid droplets in the stratosphere. Such aerosols stay in the stratosphere for several years after the eruption and reduce the insolation to the troposphere. Lowering of atmospheric temperature may follow the explosive eruption, lasting for a few years (LAMB, 1970; STOTHERS, 1984; RAMPINO and SELF, 1982; RAMPINO, 1988; SIGURDSSON, 1990). The amount of volcanic sulfur added to the atmosphere by a volcanic eruption is important to assess its effect on the earth's atmospheric environment.

The amounts of volatile emission by explosive volcanic activity in the past have been estimated by two methods; petrologic method (DEVINE et al.,1984; PALAIS and SIGURDSSON, 1989; KOHNO, 1992) and chemical method based on hydrogen ion concentration of ice core under the assumption which the global distribution pattern of the volcanic sulfuric acid aerosols and that of bomb-produced total B activity are the same (HAMMER, 1977, HAMMER et al., 1980; CLAUSEN and HAMMER, 1988; LANGWAY et al., 1988). Global acid fallout by the past large volcanic eruptions in the northern hemisphere was estimated from the analysis of acidity of Greenlandic ice core samples from Crete (71°12'N, 37°32'W) for the last 1500 years and Camp Century (71°17'N, 61°13'W) for the 10000 years (HAMMER et al., 1980). For example, the amounts of sulfuric acid emitted by volcanic eruptions of Laki (1783 A. D., Iceland, 64°42'N, 17°33'W) and of Tambora (1815A. D., Indonesia, 8°25'S, 118°00'E) were both estimated to be ca. 300 Mt (CLAUSEN and HAMMER, 1988). The ice cores from Dome C (74°40'S, 125°10'E) and Vostok (78°28'S, 106°48'E) in East Antarctica, Siple (75°55'S, 85°55'W) in West Antarctica and South Pole contain the records of several major volcanic events that took place in the southern hemisphere in the last 220 years. They are Tambora 1815,

Krakatau 1883 (Indonesia) and Agung 1963 (Indonesia) eruptions with the relative proportion of acid emission by these eruptions being approximately 5:1:1 (LEGRAND and DELMAS, 1987). Prominent bipolar excess H<sub>2</sub>SO<sub>4</sub> signals found in ice cores from Byrd (80°01'S, 119°3'W) Station and South Pole in Antarctica and from Crete and Dye 3 (65°18'N, 43°49'W) in Greenland last 1000 years were recorded in the layers whose ages were 1885, 1836, 1816, 1810, 1459, 1346, 1287, 1278, 1259, 1228 and 1168 A. D. The excess H<sub>2</sub>SO<sub>4</sub> signals in the layers dated in 1885, 1836 and 1816 A. D. were identified to originate from volcanic eruptions of Krakatau 1883, Coseguina 1835 (Nicaragua) and Tambora 1815. The other signals were regarded to be due to the aerosols released from volcanoes in low latitudes (LANGWAY et al., 1995).

An ice core was drilled at site H15 (69°04'46" S, 40°46'54" E, 1050 m a.s.l.) to a depth of 120.2 m using an electromechanical drill by the 32nd Japanese Antarctic Research Expedition in September to October, 1991 (FUJII et al., 1992, Fig.1). The electrical conductivity measurement (ECM), a measure of acidity of ice core, was made at Syowa Station soon after the core recovery. It indicates that two remarkably high peaks first appear at depths between 45 and 50 m. We have determined the chemical composition and oxygen isotopic variations of the ice core samples between for 43 and 50 m to find the causes for the high ECM peaks. We also attempted to date the ice core in order to obtain chronological information of the upper part of this core using two independent methods: (1) counting the number of peaks resulting from seasonal variations of SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup>, and (2) an empirical model of firn densification of HERRON and LANGWAY (1980).

#### 2. H15 ice core

The site H15 is located near the Soya coast, Dronning Maud Land, East Antarctica (Fig.1), where a high annual snow accumulation gives a high time resolution of the core. It is situated along the route S-H-Z from Syowa Station to Mizuho Plateau.

Table 1 summarizes the surface mass balance of snow observed at site H15 (SHIMIZU, 1975; FUJII, 1979; WADA et al., 1981; KOBAYASHI, 1982; SATOW et al., 1983; NAKAWO et al., 1984; NISHIO, 1984; FuJII et al., 1986; NISHIO et al. 1986; AGETA et al. 1987; NISHIO et al., 1988; NISHIO and OHMAE, 1989; WATANABE et al., 1990; FUJII et al., 1992; MOTOYAMA et al., 1995). The observation has been continued since April 1971. The 1972-76, 1979 and 1992 data are lacking. The mean annual accumulation rate since 1971 is 32.0 **7** 10.8 g/cm2/year. In this period, snow accumulated every year without hiatus. The mean annual accumulation rate of snow there suggests that the whole core represents the records of the last 300 years.

Bulk density of the core was measured and ice structures such as the melt layers were described soon after recovery at the Syowa Station (FUJII, 1992). The site H15 is located above the dry snow line. The mean annual atmospheric temperature is -21°C. However, ten distinct melt layers, which indicates presence of the hiatus of snow deposition, thicker than 3 mm were observed in the snow and firn layers down to 60 m below the surface. Three thick melt layers out of the 10 melt layers were observed at depths of 15.3 m, 19.1 m and 26.0 m (water equivalent depth), their thickness being 8 mm, 7 mm and 25 mm (water equivalent), respectively. The melt layers were formed during warm summer periods in the past. Existence of the melt layers may suggest that the atmospheric temperature was warmer than that at present.

#### 3. Sample preparation and analytical procedures

The core, 70 mm in diameter, was cut into half along the direction of depth after recovery, and has been preserved in a low-temperature room (-20°C) at National Institute of Polar Research. The core was sliced into a wafer (7 ~ 10 cm thick) which covers a period of 4 ~ 6 months. Surface 5 ~10 mm of the ice core was removed using a grater in order to eliminate contaminants on the surface.

Water samples after melting the ice wafers were used for measurements of the pH, major chemical composition and oxygen isotopic ratios (<sup>18</sup>O/<sup>16</sup>O ratio). The core samples were melted in a microwave oven in a Teflon beaker that had been cleaned with ultra-pure water, and filtered with a 0.2 mm filter paper (Nuclepore polycarbonate membrane, Corning Costar Corp.) in a clean room (class-1000) at the National Institute of Polar Research. The residue on filter paper was preserved for observation of solid residue.

Acidity, which is an index of acidic aerosols accumulated with snow, was measured using a pH meter (HM-60S, TOA Electronics Ltd.). The pH meter consists of a glass electrode separated from a

comparative reference electrode (FAR-101, TOA Electronics Ltd.), because melt water of ice cores had a small buffer capacity due to very low ionic concentration (FUJII et al., 1989). Concentrations of major anions (Cl-, NO<sub>3</sub>- and SO<sub>4</sub><sup>2-</sup>) and cations (Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup> and Ca<sup>2+</sup>) were measured with ion chromatography using a Dionex ion chromatograph (model 2000i) equipped with HIPC-AG4A and -CG3 concentrators, AS4A and CS3 separator columns, and anion and cation micro membrane suppressors to concentrate the ionic species that are dissolved in a ~100 ppb level. Replicate measurements of standard solutions prepared from analytical grade reagents (NaCl, NaNO<sub>3</sub>, K<sub>2</sub>SO<sub>4</sub> and KCl, Mg and Ca 1000ppm standard solutions) indicated the relative analytical error of 5% of a measured value. The <sup>18</sup>O/<sup>16</sup>O ratio (<sup>18</sup>O) of samples was measured with a stable isotope mass spectrometer (MAT-262) using the H<sub>2</sub>O-CO<sub>2</sub> exchange technique by EPSTEIN and MAYEDA

(1957), and reported in a conventional  $^{18}$ O expression with respect to SMOW.

### 4. Dating of the H15 core

Dating methods of ice cores proposed so far include (1) the stake method (annual accumulation rate), (2) the empirical model for firn densification (HERRON and LANGWAY, 1980), (3) the stratigraphical method based on seasonal variations of <sup>18</sup>O, particle concentration, chemical composition, pH and ECM profiles (EPSTETN et al., 1965; HAMMER et al., 1978; FUJII, 1981; THOMPSON et al., 1981; LEGRAND and DELMAS, 1984; WATANABE et al., 1988), (4) detection of key volcanic signals such as tephra and high SO<sub>4</sub><sup>2-</sup> concentration from volcanic eruption of a

known date (HAMMER, 1977., HAMMER et al, 1980), and (5) the  $^{210}$ Pb method (CROZAZ and LANGWAY, 1966; MASUDA and HARADA, 1986). Since the mean annual accumulation rate of snow at site H15 from 1971 to 1993 varies as much as  $\pm 30$  % (Table.1), it is difficult to use the mean accumulation rate for accurate dating of the H15 core. In this study, we thus adopted the stratigraphical method, i.e., counting the number of ECM peaks due to seasonal variations in snow chemistry, and the snow densification model proposed by HERRON and LANGWAY (1980) for the core dating.

Concetration of excess sulfate (or non sea-salt sulfate) of aerosols in the atmosphere at Mawson (67°36'S, 62°53'E) in East Antarctica is highest in January and relatively low from May to September (PROSPERO et al., 1991). The atmospheric excess  $SO_4^{2-}$  concentration is governed by the oxidation of biogenic dimethylsulfide (DMS) to  $SO_2$  and methanestulfonic acid. The extent of oxidation varies with atmospheric temperature (HYNES et al., 1986, KOGA et al., 1990). The sulfur dioxide of DMS origin is oxidized to form sulfuric acid in the atmosphere (HATAKEYAMA et al., 1985) and then accumulates in snow by wash-out. Seasonal variations of  $SO_4^{2-}$  and  $NO_3^{-}$  concentrations in drifting snow which collected at Mizuho Station during February to December in 1986 increased significantly in summer (OSADA et al., 1989; OSADA, 1994), and the pH values of drifting snow samples indicate that snow in summer is more acidic than that in the other seasons (FUJII, 1989). Therefore, it is conceivable that an ECM peak reflects summer acidity of snow, and that ice core can be dated by counting the ECM peaks.

When DC voltage is applied across a piece of ice through electrodes, a weak electric current is induced. The current depends on the concentration of mobile ions, their mobility and efficiency of charge exchange at the electrodes. Since the mobility of  $H_3O^+$  ion is much greater than that of OH-

and other ions, the initial current is proportional to the  $H_3O^+$  concentration or acidity. This is the principle of ECM (HAMMER, 1980; HAMMER, 1983). Figure 2 shows the ECM profile down to 50 m. The core depth has been converted to the water equivalent depth based on the density profile shown in Fig.3. The profile represents the ECM values after taking a running average of every 10 cm interval. Above 50 m, we can find two high ECM peaks at depths of 46.7 m and 48.3 m (Fig.2).

Figure.4a compares the ECM profile of the top 4 m of the H15 core with the dates of firn given by the stake observation (Table 1). We have assigned high ECM peaks to summer of an appropriate year based on the stake observation. Summer peaks are indicated by a number by a high ECM peak, although there are a few exceptions. It should be noted that the highest ECM peak is found between the two ECM peaks assigned to 1983 and 1984 summers. Excess  $SO_4^2$ - concentrations shown in

Fig.4b were calculated using the following equation assuming that Na<sup>+</sup> ion was entirely derived from seaspray.

$$(SO_4^{2-})_{excess} = (SO_4^{2-}) - (SO_4^{2-} / Na^+)_{seawater} (Na^+)$$
 (1)

where  $(SO_4^{2-})$  and  $(Na^+)$  are the observed  $SO_4^{2-}$  and  $Na^+$  concentrations of ice core samples (Table 2) and  $((SO_4^{2-}/Na^+)_{seawater}$  is the ratio of  $SO_4^{2-}$  to  $Na^+$  concentrations of seawater that is 0.25 (weight ratio). The excess  $SO_4^{2-}$  concentration generally varies between 0 and 70 ppb (Fig.4b). High excess  $SO_4^{2-}$  peaks roughly agree stratigraphically with high ECM summer peaks. However, there are a few sporadic negative peaks (not shown in Fig.4b), which suggests existence of excess  $Na^+$  of unknown reasons. It should be noted that there is a layer that contains very high excess  $SO_4^{2-}$  concentration at

2.3 m depth. Deposition of this excess  $SO_4^{2-}$  occurred in the latter half of 1983, and is likely to reflect deposition of sulfuric acid aerosols originated from an explosive volcanic eruption of El Chichon 1982 (Mexico, 17°36'N, 93°23'W) as will be discussed later. From the foregoing, counting the summer peaks in the ECM profile is certainly usable to date the ice core.

The high ECM peaks shown in Fig. 2 were counted from the surface to 50 m depth. Above 50m, there are two remarkably high peaks at 46.7 m and 48.3 m depths. They correspond to~1826 A. D. (~165 B. P.) and -1821 A. D. (~170 B. P.), respectively, although this dating method may involve errors due to over-counting of non-summer ECM peaks such as a volcanic signal at 2.3 m depth (EI Chichon 1982) or under-counting of small summer peaks and undetected summer peaks caused by melting of snow as observed at depths of 15.3 m, 19.1 m and 26.0 m (water equivalent depth).

An empirical model of firn densification was proposed by HERRON and LANGWAY (1980). This model is based on the principle that air space of the firn is linearly related to stresses resulting from weights of overlying snow. According to formulation by HERRON and LANGWAY (1980), there is a linear relationship between depth of firn and in ( $/(i^-)$ ), where i is a constant density of ice (0.917 g/cm3) and r is the density of the firn layer at a given depth. The plot of the ln( $/(i^-)$ ) versus depth relationship is divided into two segments based on the densification rate. The first segment is for < 0.55 g/cm3 and the second segment is for  $0.55 < < 0.82 \sim 0.84$  g/cm<sup>3</sup>. These segments correspond to the stages where settling and packing of snow grains and further compaction process take place. Since "pore close-off" will occur above the densities greater than  $0.82 \sim 0.84$  g/cm<sup>3</sup>, the above model cannot be applied for ice whose density is greater than 0.84 g/cm<sup>3</sup>. The relationship between ln( $/(i^-)$ ) and raw depth that has been converted from the density profile (Fig. 3) is shown in Fig.5. HERRON and LANGWAY (1980) related the annual accumulation rate A to the slope C of the ln( $/(i^-)$ ) versus depth relationship as,

 $A = (i \cdot k_1 / C)^2$  (2)

where  $k_1$  is a constant relating to the densification rate. The value of  $k_1$  is a function of the mean atmospheric temperature and is given in HERRON and LANGWAY (1980). Taking the mean atmospheric temperature of -21°C at site H15 and the slope C of regression lines of the first and second stages in Fig. 5 to be 6.5 x 10<sup>-2</sup> and 3.8 x 10<sup>-2</sup>, respectively, the accumulation rate A is calculated to be 145 and 51 cm/year for the first and second stages, respectively. The time necessary for snow to attain the maximum density of the first stage( = 0.55 g/cm<sup>3</sup>; 6.2 m in raw depth or 2.9 m in water equivalent depth) is calculated to be 2 years by dividing 2.9 m by the accumulation rate. Indeed, the density of 0.55 g/cm3 is found at 7.8 m (Fig. 3). Similarly, the time required to attain the maximum density of the second stage ( = 0.84 g/cm<sup>3</sup>; 58.8 m in raw depth or 42.9 m in water equivalent depth) is calculated to be 148 ± 12 years. The bottom of firn in the second stage thus deposited ~150 years ago, corresponding to 1841 ± 12 A. D. The error arises from the regression of the ln( /(  $_{i}$ - )) versus depth relationship. As mentioned earlier, there are two high ECM peaks at 46.8 and 48.3 m (water equivalent depth, Fig.2). The dates of these peaks should, therefore, be older than 1841 ± 12 A. D., and are roughly estimated to correspond to 1822~1828 (±12) A. D., assuming a constant accumulation rate below the layer with = 0.84 g/cm<sup>3</sup>.

## 5. Acidity, SO<sub>4</sub><sup>2-</sup> concentration and <sup>18</sup>O measurements

Acidity, chemical composition and <sup>18</sup>O of the H15 core samples were measured from surface to 50 m depth. As stated previously, the ECM and excess  $SO_4^{2-}$  peaks at 2.3 m depth are present between 1983 and 1984 (Fig. 4) and are likely to be related to deposition of stratospheric aerosols released by the volcanic eruption of El Chichon 1982. Other possible explosive volcanic activities prior to 1984 are the eruptions of Galunggung 1982 (Java, 7°03'N, 108°01') and Colo 1983 (Indonesia, 0°02'N, 121°06'E) whose Volcanic Explosivity Index (VEI) are 4(SIMKIN and SIEBERT, 1994). Of these explosive eruptions, the El Chichon eruption of 1982 in April (VEI=5) is most likely to be responsible for the high peaks because of its large magnitude. Its maximum column height is estimated to have reached 32 km (CAREY and SIGURDSSON, 1982), high enough to give a global impact. The eruption injected a large amount of sulfur to the stratosphere, an estimate being 3.3 ~ 10 Mt by the total ozone mapping spectrometer on the Nimbus 7 satellite (EVANCE and KERR, 1983; KRUEGER, 1983). The aerosols spread over the globe at the latitude of 20°N for about 20 days and further to higher latitude following the general atmospheric circulation. The solar radiation into the atmosphere observed at the Syowa Station started to decrease from 1982 in November, and was obviously low in January, 1983 (YAMAUCHI and SHIMURA, 1984). A time lag of over 1 year between the eruption (1982 in April) and the date (mid-1983) when the high excess SO<sub>4</sub><sup>2-</sup> peak was observed in Fig.4b is explained by a long residence time of volcanic aerosols in the stratosphere. This is consistent with the fact that the stratospheric aerosols was present for about 2 years after the explosive volcanic eruption of Mt. Pinatubo (Philippines, 15°13'N, 120°35' E) in June 1991 (McCORMIC et al., 1995). The PS1 and PS14 ice cores drilled at South Pole in 1983 ~ 1984 summers do not contain a prominent volcanic signal of the 1982 El Chichon eruption (DELMAS et al., 1992). It may have taken longer for the volcanic aerosols to reach South Pole compared to the site of H15 since the latitude of South Pole is higher than that of H15, and the aerosols may not have deposited yet at South Pole where the ice cores were drilled in 1983 ~ 1984 summers.

The H<sup>+</sup>, excess SO<sub>4</sub><sup>2-</sup> and <sup>18</sup>O variations were measured in detail for depths from 43 m to 50 m in order to constrain the causes for the two high ECM peaks at 46.7 and 48.3 m depths (Fig. 2). The results are shown in Table 3 and Fig. 6. It is obvious that the two high ECM peaks are accompanied by H<sup>+</sup> (Fig. 6b) and excess SO<sub>4</sub><sup>2-</sup> peaks (Fig. 6c). The excess SO<sub>4</sub><sup>2-</sup> concentrations of 350 ng/g at 46.7 m and 200 ng/g at 48.3 m correspond to 7 and 4  $\mu$  equiv. H+/kg, respectively, assuming that the excess SO<sub>4</sub><sup>2-</sup> deposited as sulfuric acid. These values are in reasonable agreement with the directly measured  $H^+$  concentrations at corresponding depths (Fig. 6b). This is a good indication that excess  $SO_4^{2-}$ originates from acid fallout as sulfuric acid. From the discussion of dating, the layers containing these high acidity peaks correspond to the beginning of 19th century. Two significant sulfate peaks have been found in the ice core drilled at Stations Dome C and Vostok (LEGRAND and DELMAS, 1987) in East Antarctica, Siple (DAI et al., 1991) and Byrd (LANGWAY et al., 1994; LANGWAY et al., 1995) in West Antarctica and South Pole (DELMAS et al, 1992, LANGWAY et al., 1995). The two high SO<sub>4</sub><sup>2-</sup> peaks in the Dome C core are assigned to have originated from the unknown 1809 eruption and Tambora 1815 eruption (MOORE et al., 1991; DAI et al., 1991; DELMAS et al., 1992; LANGWAY et al., 1995), High SO<sub>4</sub><sup>2-</sup> layers are observed in ice cores of a similar age range at Siple and Byrd in West Antarctica and South Pole and Site A (70°38'N, 35°49'W), Site T (72°35'N, 38°27'W), Dye 3 and Crete in Greenland. A comparison of SO<sub>4</sub><sup>2</sup>-concentrations of these cores indicates that an explosive volcanic eruption occurred in 1809 or earlier in the low latitude and the amount of stratospheric aerosol emission was about half that of Tambora 1815, although a volcanic event in 1809 has not been described in the ancient literature (DAI et al., 1991; LANGWAY et al.,

1995). Of these cores, hydrogen ion concentration of a layer containing aerosols from the Tambora eruption is higher than that from the 1809 unknown eruption by a factor of  $1.5 \sim 2$  except at Dome C (Legrand and Delmas. 1987; Moore et al., 1991; Dai et al., 1991; Delmas et al., 1992; Langway et al., 1995). At H15, the total amounts of H<sub>2</sub>SO<sub>4</sub> deposition were calculated by integrating the areas of the

H<sup>+</sup> peaks at 46.7 m from 46.46 to 46.90 m and at 48.3 m from 48.12 to 48.27 m, respectively (Fig. 6). The amounts of H<sub>2</sub>SO<sub>4</sub> were calculated to be 68 kg/km<sup>2</sup> at 46.7 m and 44 kg/km<sup>2</sup> at 48.3 m with the ratio of 1.5. This fact, together with the observations at the other sites that the Tambora 1815 eruption emitted more acid than the unknown 1809 cruption by a factor of  $1.5 \sim 2$ , suggests strongly that the 46.7 m peak corresponds to the Tambora 1815 eruption and the 48.3 m peak corresponds to the unknown 1809 eruption.

The layers at 46.7 m and 48.3 m depths were initially dated to be 1828 and 1822 A. D., respectively, by the empirical densification model (HERRON and LANGWAY, 1980), and to be 1826 and 1821 A.D., respectively, by the counting the ECM highs. The difference of ~10 years may be ascribed to the presence of unconformity as indicated by distinct three melt layers between 15 m and 26 m (water equivalent depth). MACHIDA et al (1995) assumed that the ECM peaks at 46.7 m and 48.3 m depths correspond to the Tambora 1815 and unknown 1809 eruptions in their estimation of the annual accumulation rate of 26.0 g/cm<sup>2</sup>/year. The present study supports their assumption.

Andesitic fine ash was found in layers dated to 1809-1810 and 1816 A. D. in South Pole and Dome C cores (DE ANGELIS et al., 1985; PALAIS et al., 1990). Fine ash-like materials (~10 mm) have been found in the corresponding layers of H15 core. If the materials are proven to be of volcanic origin through chemical analysis, our interpretation will be further strengthened.

Except the high ECM peaks at 46.7 and 48.3 m, we can see ECM peaks that are significantly higher than the background variations (Fig. 2). Sulfate concentrations were measured for samples of high ECMs (> 2  $\mu$ A above 15 m depth and >4  $\mu$ A below 15 m depth). If the SO<sub>4</sub><sup>2-</sup> concentration is higher than 100 ng/g (note that the background SO<sub>4</sub><sup>2-</sup> concentration is about 50 ng/g, see Fig. 4) and if the layer is thicker than 0.1 m (water equivalent), we regard such peaks to represent volcanic signals. There are altogether eight ECM highs satisfying the above conditions. Table 4 compiles these eight

ECM highs in the H15 ice core, together with the estimated dates, maximum  $SO_4^{2-}$  concentration, and

duration of ECM highs. Duration was calculated from the thickness of a SO<sub>4</sub><sup>2-</sup> concentration peak and the annual accumulation rate of 26.0 g/cm2/year by MACHIDA et al. (1995). We attempted to relate those ECM highs to possible volcanic eruptions. After 1815, two volcanic eruptions are known to have the VELs greater than 5 (SIMKIN and SIEBERT, 1994), and therefore tp give large aerosol loading to the atmosphere; Coseguina 1835 (Nicaragua, 12°09'N, 87°05'S) and Krakatau 1883 (Indonesia, 6°01'S, 105°04'E, RAMPINO and SELF, 1982; DEVINE et al., 1984; SELF et al., 1989). The volcanic signals due to these eruptions have been found in the ice cores from Dome C, Vostok, Siple, Byrd and South Pole (LEGRAND and DELMANS, 1987; DAI et al., 1991; DELMAS et al., 1992; LANGWAY et al., 1994; LANGWAY et al., 1995). The ECM highs that are likely due to these eruptions occur at the water equivalent depths of 39.57 and 26.29 m in the H15 core, and are dated to be 1839 and 1890, respectively, based on the stake method using the annual accumulation rate of 26.0 g/cm<sup>2</sup>/year by MACHIDA et al. (1995). These estimated dates are younger than the actual dates of respective eruptions by 5-7 years. This probably arises from the formation of melt layers as will be discussed later. Since there is no known big volcanic cruption around 1859 which corresponds to 34.4 m in the H15 core, the causes of this ECM high is not clear. The ECM highs at 10.99 m (estimated to be 1949 A.D.) and at 9.09 m (estimated to be 1959 A.D.) may have resulted from the volcanic eruptions of Nilahue 1955 (Chile, 40°04'S, 72°01'W) and Agung 1963 (Indonesia, 8°03'S, 115°05'E), respectively. The difference of 6 years between the actual and estimated dates probably reflects the "unconformity" effect as mentioned below.

In Table 4 were listed the estimated dates of the ECM (and  $SO_4^{2-}$ ) highs and the dates of the volcanic eruptions that might have caused the ECM highs. It is known that there are three thick melt layers at 15.3, 19.1 and 26.0 m (water equivalent depth), which indicates presence of the hiatus of snow deposition. If the hiatus did occur, the accumulation rate of 26.0 g/cm<sup>2</sup> by MACHIDA et al. (1995) becomes an underestimate. With this value, the layers above the melt-containing layers would be dated younger than the actual dates, whereas the layers below would be dated older, since the period for the hiatus would have been included in the calculation of dating. The difference between the

estimated and actual dates should thus be negative below 26.0 m and positive above this depth. This is quite consistent with the above comsideration of the effect of the hiatus on the accumulation rates. Different mean accumulation rates must be used below and above the melt layers.

The <sup>18</sup>O values of rain and snow vary with the average localatmospheric temperature (DANSGAARD, 1964). Consequently, seasonal changes of atmospheric temperature and long-term climatic changes are reflected in the oxygen isotopic variations of ice cores (JOHNSEN et al., 1972; HAMMER et al., 1980; DAI et al., 1991; DANSGAARD et al., 1993; GREENLAND ICE-CORE PROJECT MEMBERS, 1993). Figure 6d shows the <sup>18</sup>O variations between 43 and 50 m depths of the core. Unfortunately, we cannot recognize the summer highs and winter-lows in the <sup>18</sup>O variations. There is a tendency that the <sup>18</sup>O values decreased after rhe eruption of the Tambora 1815 eruption (46.7 m depth), but it is not the case for the <sup>18</sup>O values after the 1809 unknown eruption (48.3 m depth). The <sup>18</sup>O profile of Siple core indicated that short-term temperature changes after explosive volcanic eruptions such as Tambora 1815 were not recorded in the core (DAI et al., 1991). Thus, interpretation of the <sup>18</sup>O variations as volcanic signals is not straightforward.

## 6. Conclusions

Two remarkably high peaks of ECM, acidity and excess SO<sub>4</sub><sup>2-</sup> concentrations were found at the water equivalent depths of 46.7 m and 48.3 m of the 120-m ice core drilled at site H15 in East Antarctica. These signals are considered to have originated from the volcanic eruptions of Tambora in 1815 A. D. and unknown in 1809 A. D. The dates were estimated from two independent methods by counting the number of ECM peaks and an empirical model of firn densification. High SO<sub>4</sub><sup>2-</sup> peak detected at depth of 2.3 m most likely resulted from a volcanic eruption of El Chichon 1982. There is a possibility that the ECM peaks associated with SO<sub>4</sub><sup>2-</sup> highs between Tambora 1815 and El Chichon 1982 eruptions may represent the effect by the volcanic eruptions of Coseguina 1835, Krakatau 1883 and Agung 1963.

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Period	Annual accu	mulation rate
	Snow depth*	Water equiv. depth**
Ang 1071 Ang 1072	chu yr	g/cm= /yr
Apr. 1971 - Apr. 1972	51.0	21.0
Jan. 1977 - Pcb. 1978	127.0	55.8
Jan. 1978 - Jan. 1979	72.0	30.5
Jan. 1980 - Jan. 1981	63.0	26.7
Jan. 1981 - Jan. 1982	108.0	45.8
Jan. 1982 - Jan. 1983	87.5	37.1
Jan. 1983 - Dec. 1983	74.3	31.5
Dec. 1983 - Jan. 1985	60.0	25.4
Jan. 1985 - Jan. 1986	89.0	37.7
Jan. 1986 - Jan. 1987	89.0	37.7
Jan. 1987 - Jan. 1988	25.0	10.6
Jan. 1988 - Dec. 1988	35.0	14.8
Dec. 1988 - Jan. 1990	75.0	31.8
Jan. 1990 - Jan. 1991	87.0	36.9
Jan. 1991 - Oct. 1991	78.5	33.3
Jan. 1993 - Jan. 1994	87.0	36.9
Average	75.5	32.0
Std dev.	25.4	10.8

Table 1.	The annual accumulation rate of snow
	between 1971 and 1991.

\* Data compiled from JARE data report (1972-1995).
 \*\* Accumulation rate of water was calculated using mean surface snow density of 0.42 g/cm<sup>3</sup> measured along Route S 29 to H 54 (Nishio, 1984).

Water equiv. depth	H*	Cl+	NO3"	SO42-	Na <sup>+</sup>	K*	Mg <sup>2+</sup>	Ca <sup>2+</sup>	δ18Ο
m	n mol/g	ppb	ppb	ppb	ppb	ppb	ppb	ppb	700
0.02	3.09	318	0	65	183	5	101.		-24.7
0.05	3.72	301	11	55	170	5	•		-24.8
0.10	2.75		0	56	187	8			-27.3
0.14	3.31	566	0	106	322	17		12.00	-25.1
0.20	4.47	344	21	82	195	14	•		-22.9
0.35	5.62	178	93	90	93	3	R		-21.7
0.39	4.68	721	56	111	387	15			-25.8
0.43	3.98	884	14	83	505	20		•	-29.8
0.47	3.98	298	22	36	104	2		1.1.	-25.1
0.51	3.89	309	12	30	109	2			-24.7
0.55	3.98	192	13	19	65	0	•		-24.2
0.60	3.63	404	13	29	119	2			-26.3
0.64	3.80	656	10	121	343	8			-20.6
0.69	3.63	453	8	56	156	0			-20
0.74	3.72	628	8	85	253	6			-19.5
0.78	3.89	89	30	53	48	0			-20.4
0.83	437	47	32	48	31	0			-23.5
0.87	3.55	87	1	34	44	0			-27.2
0.07	3.89	294	16	28	96	0			-29.9
0.92	3.63	299	1	32	101	0			-27.8
1.01	3.98	350	ō	55	124	0			-24
1.06	3.47	78	27	35					-23.8
1.00	3.72	10				-			-22.1
1.11	3.47								-23.1
1.10	3.47	330	22	66					-26.8
1.21	3.80	220	40	41	103	2	12	9	-31.7
1.20	3.55	180	23	32	95	2	9	8	-29.6
1.51	2.55	146	58	36	81	1	9	15	-23.6
1.57	2.00	238	14	63	144	26	15	12	-20.9
1.43	3.33	212	47	101	116	2	12	6	-21.2
1.48	4.00	60	24	32	35	ã	4	5	-22.8
1.54	3.03	140	10	41	55				-22.7
1.60	3.72	140	10	41	61.15				-23 1
1.65	3.09							1.22	-23 5
1.70	3.12			24	1				.24 3
1.75	3.24	211		24		100			-24 5
1.80	3.09		-	67	846		10		-243
1.85	2.88	231	:	42	040		6		.25 5
1.89	3.16	122	2	42	010	5	0		-20.0
1.94	3.89	256	14	100					.28 1
1.97	3.02	419	4	19		•		-	-40.1

Table 2. H<sup>+</sup> concentration, chemical composition and  $\delta^{18}O$  from surface to 4 m depth of H15 core.

Tabl	e 2.	(continued)
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Water equiv. depth	H+	Cl+	NO3	SO42-	Na <sup>+</sup>	K*	Mg <sup>2+</sup>	Ca2+	δ <sup>18</sup> O
m	n mol/g	ppb	ppb	ppb	ppb	ppb	ppb	ppb	%0
2.00	2.88	380	13	60			-	outing and	-27.6
2.06	3.16	1.5.							-28
2.11	2.95								-27.5
2.15	3.02	107	9	25					-25.1
2.21	3.24					-			-22
2.26	3.24	2452	68	180					-18.3
2.31	3.09	2442	17	438					
2.34	3.55	104	9	99	813		6		-22.1
2.37	3.02		14	490					-21.7
2.41	3.39	603	7	104					-20.3
2.46	1.26	-							-20.4
2.52	2.57								-21
2.57	3.02	170	5	52			-		-20.3
2.62	3.16								-20.8
2.66	3.72	85	29	89	815		3	15	-22.6
2.70	3.24	96	21	86	908		4	23	-26.1
2.75	3.24	205	16	73					
2.81	2.95	224	13	60			-		-23.4
2.87	3.31				-				-20.9
2.93	3.55		÷ * 1						-21.3
2.98	3.55	161	11	30	89	2	10	6	-24.3
3.04	3.55	99	6	22	60	1	6	4	-26.2
3.10	3.80	98	30	30	59	1	6	7	-25.5
3.15	3.39	134	33	24	78	1	8	8	-22.8
	3.80	262	4	44	158	2	17	8	-19.9
3.27	3.16	330	4	56	184	4	20	8	-19.2
3.33	3.39	170	6	36	98	2	11	5	-19.2
3.39	3.39	51	12	26	28	0	4	3	-19.4
3.45	3.63	132	21	57					-21.1
3.51	3.39								-24.2
3.56	3.31	-		-	100 - 17 C	•			-27.9
3.62	3.55	213	34	44					-27.1
3.69	3.47								-23.3
3.75	3.31				-				-22.3
3.82	3.55	220	11	43					-21.6
3.87	3.31	147	1	59	144		10	7	-21.6
3.90	3.63	187	14	68	175		12	8	-
3.95	3.39	195	7	28					-21.9

- : not determined.

m 42.98 43.07 43.16	n mol/g	ppb	mh	mh					~
42.98 43.07 43.16	2.95	ppo		1000	DDD	DDD	ppp	ppb	100
43.07 43.16		78	6	15	37	. 0	3	. 1	-24.3
43.16	3.55	168	20	38	91	2	7	2	-23.3
45.10	3.00	133	10	16	70	ō	5	1	-23.1
42.25	3.80	155							
43.23	3.89						101		
43.54	3.89								
43.43	3.80								
43.52	4.17								
43.00	4.17	162	22	22	07	0		1.1.1	.264
43.08	3.31	100	21	63	324	10	1.1.24		.26.9
43.70	3.31	201	26	45	158	3			-26.4
43.84	3.33	201	20	40	150	0	1.0.0	1	.24 3
43.91	3.47	30	30	12	20				226
43.98	3.31	66	20	1/	39	2			-22.0
44.05	3.63	199	14	45	129	3			-21.0
44.12	3.63	497	10	115	295	0			-22.0
44.19	3.72	126	39	00	10	0	0.15		-23.7
44.27	3.47	224	14	25	110	4			-24.9
44.35	3.39	523	18	95	18	0			-24.9
44.42	3.72	358	25	98	211	4		-	-24.7
44.49	3.55	108	22	47	70	0			-24.1
44.56	3.55	101	24	22	45	0			-24.4
44.63	3.72	17	55	46	6	0		•	-23.4
44.71	3.24	183	18	35	102	0	2.4.4.5		-23.9
44.78	3.09	356	13	71	171	0			-24.8
44.85	3.72	64	31	71	35	0			-24.4
44.92	3.89	147	40	87	73	0	•		-23.4
44.99	3.39	136	29	54	80	0			-22.8
45.07	2.95	63	12	23	36	0			-22.9
45.14	2.95	68	13	19	38	•		•	-22.1
45.22	4.68	77	14	47	40	0	3	1	-21.5
45.31	4.90	109	19	70	59	0	6	2	-21.6
45.40	4.79	50	25	50		0	2	1	-22.6
45.49	3.09	381	16	68	208		19	10	-24.3
45.57	3.09	152	17	40	34		14	8	-24.2
45.65 -	3.89								
45.74	3.72					•			
45.82	3.80				-				
45.90	3.55							-	
45.99	3.47						5		•
46.06	4.47	108	30	53	137		5	3	-23.4
46.13	5.13	208	52	104	114		15	8	-23.3
46 20	4.57	118	26	52	66		7	3	-23.3
46.27	4.27	176	14	49	90		12	4	-22.7
46 33	5.25	254	23	94	92	-	44	4	-23.6
46.40	417	1413	32	114	866	-	34	39	-25.2
46.46	5.62	328	17	167	255		38	6	-27.2
46.50	575	130	47	86	146		6	3	-27.2
40.52	537	424	22	147	240		34	10	-25.2
40.50	5.80	505	32	207	368		37	13	-23.4

Table 3. H<sup>+</sup> concentration, chemical composition and  $\delta^{18}O$  at a depth between 43 and 50 m of H15 core.

# Table 3. (continued)

Water equiv. depth	H*	Cl*	NO3	SO42-	Na <sup>+</sup>	K*	Mg <sup>2+</sup>	Ca <sup>2+</sup>	ô18O
m	n mol/g	ppb	ppb	ppb	ppb	ppb	ppb	ppb	%0
46.70	10.47	166	68	397	167		6	4	-22.4
46.76	7.59	84	40	257	45		6	2	-21.9
46.83	5.37	174	9	108	85		11	3	-22.1
46.90	5.37	217	16	163	219		15	6	-22.8
46.96	4.57	229	18	68	271		23	5	-23.1
47.02	4.17	1304	15	112	844		36	41	-22.7
47.09	3.63	763	9	90	409		54	17	-22.2
47.15	4.37	226	12	50	119		23	4	-21.7
47.21	4.47	432	14	112	293		29	10	-21.7
47.28	4.68	304	9	44	134		33	6	-217
47.34	4.17	1001	19	156	634		39	29	-20.5
47.40	4.17	265	8	42	225		27	5	.196
47 47	417	187	10	36	111		14	8	.10 2
47 54	437	413	10	66	222	-	24	10	100
47.60	513	357	13	100	100		20	0	10.5
47.67	4 90	164	26	62	87		10	0	-19.5
47.07	4.90	104	64	10	144		2	4	-20.0
47.74	4.90	127	26	22	77		5	0	-20.4
47.01	4.79	157	20	20		1.	2	4	-20.8
47.07	4.19	94	15	00	33			1.	-21.5
47.93	5.04	114	44	107	14	1.1.1	81.4X.M		-23.0
48.00	0.51	10/	49	10/	101				-24.5
40.00	5.75	200	41	93	315				-24.6
48.12	1.59	483	35	202	120			•	-24.4
48.19	0.40	204	41	204	139			•	-23.4
48.27	7.41	158	50	243	13	-			-23.4
48.35	5.13	42	28	51	136				-23.9
48.42	5.57	81	49	33	185		•		-25.1
48.49	4.19	240	30	43	283			-	-26.4
48.57	5.13	274	16	34	208				-27.2
48.64	5.50	174	42	11	251		1.1.2		-24.2
48.71	6.03	203	43	139	92				-23.2
48.80	3.72	197	12	32	112	1	12	4	-25.3
48.89	3.63	229	13	42	132	3	11	3	-24.1
48.97	4.37	396	17	75	216	5	8	3	-23.4
49.04	4.90	200	20	83	96	1	23	4	-22.7
49.12	4.57	525	26	72	289	6	17	12	-22.6
49.22	5.01		-						
49.33	4.27								
49.43	4.07								-
49.52	3.89		ST-04						
. 49.60	3.55	-	-						-
49.70	3.39							-	-0
49.79	3.39	-						-	1.
49.88	3.89	-	-				-	-	-
49.98	3.55	2.		100		1.	1. 1.		
· not determined				2 sector in a local			Charles and a local sector		

Water equiv. depth	Estimated date	Maximum excess SO42-	Duration	Possible volcanic eruption	Δt**
m	A. D.	ng/g	year		year
2.31	- 1983	186	0.4	El Chichon 1982	-1
9.09	1957	136	1.0	Agung 1963	-6
10.99	1949	134	0.5	Nilahue 1955	-6
26.29	1890	207	0.9	Krakatau 1883	+7
34.40	1859	140	0.4	?	. ?
39.57	1840	128	0.5	Coseguina 1835	+5
46.70	1815*	355	1.9	Tambora 1815	-
48.27	1809*	225	0.9	unknown 1809	

Table 4. Eight ECM highs in the H15 core and their relation to possible volcanic eruptions.

\*Fixed date.

\*\*Difference between the actual and estimated dates.







Figure 2. ECM profile of the upper 50 m of the H15 core. The profile was obtained by taking a running average of every 10 cm interval. The core depth was converted to water equivalent depth based on the measured density of firm.



Figure 3. Density profile of the upper part of the H15 core. The density was measured for about 30 samples from firm layers and 10 samples from ice layers.



Figure 4. (a) ECM profile from the surface to 4.0 m depth (reproduced from Fig. 2). Dates of snow from the stake observation were indicated on the top scale. ECM peaks have been assigned to an appropriate dates based on the annual accumulation rate from the stake data. (b) Excess sulfate concentration profile from the surface to 4.0 m depth.



Figure 5. Relationship between ln ( $\rho/(\rho_i-\rho)$ ) and depth of the H15 core. (a) The first stage:  $\rho < 0.55$  g/cm<sup>3</sup>, (b) The second stage:  $0.55 < \rho < 0.84$  g/cm<sup>3</sup>.



Figure 6. Profiles of (a) ECM (reproduced from Fig. 2), (b) H+ concentration, (c) excess  $SO4^{2-}$  concentration and (d)  $\delta^{18}O$  of the H15 core from 43 m to 50 m depth.